

# Preview

Please note that these experiments are the **SAME** ones included in my science units listed on the next page in the Table of Contents. They can definitely be used without the units which is why I decided to offer them as a stand-alone product.

### Table of Contents

Pages	→ Unit/Topic	Experiment
3-11	Potential & Kinetic Energy	The Incline Roll
12-20		Transferring Energy
22-28	Push and Pull forces	Plastic Bottle Bowling
29-34		Incline Ramp
36-42	Effects of Heat	Heating up Glowsticks
43-49		Expanding a Liquid
51-60	Light	Refraction with Water
61-73		Opaque, Translucent, Transparent
75-81	Physical & Chemical Changes	Change in Temperature
82-88		Creating a Gas
90-98	Scientific Method	Turning a Penny Green
99-107		Balloons in the freezer
108-109	Terms of Use	



### Creating a Gas: Teacher Directions

- I like using the team format to encourage cooperation and communication. It also provides great opportunities to use each other's names and even record them as team members.
- · Hopefully you have already talked with and taught your students about lab safety. Although there is nothing dangerous in these experiments, it is important to always follow good safety rules.
- · I also really try to come up with activities using items you likely have in the classroom on at home.
- Allow students to collect the materials needed and check them off as they obtain them.
- · Make sure you review ways you can tell a chemical change has occurred. One of them is that a chemical reaction will cause gas to be released. That is what we will be observing in this activity.
- · You will be delivering the directions verbally. My students were either non-readers or emerging, so I did not have the directions here for them to read.
- Students record the results and test their hypothesis.

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### Physical or Chemical Change Experiment #2 Producing a Gas

People on my team:

Materials needed:



















Physical or Chemical Change Experiment #2 Producing a Gas

My hypothesis

I think the balloon on:

Bottle #1 will

Bottle # 2 will











Pg 3

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Physical or Chemical Change Experiment #2

Producing a Gas

### The Experiment:

### Balloon #1

- 1. Carefully place the balloon on the bottle without
- letting the contents spill in. (Teacher to do)

  2. Lift up the balloon and empty the contents into the bottle.
- 3. Observe changes in the balloon.

Balloon #2 : Repeat same process



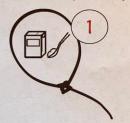


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Physical or Chemical Change Experiment #2 Producing a Gas

Preparing Materials

In each balloon, put 1 teaspoon of:





In each bottle, put 1 cup of:





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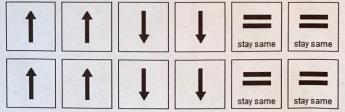
Pg 2

### HILL GOOD COLUMN

## Physical or Chemical Change Experiment #2 Producing a Gas

### Testing my hypothesis:

	Prediction of Balloon Change	Result of Balloon Change
Bottle #1		
Bottle #2		



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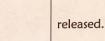
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## Physical or Chemical Change Experiment #2 Producing a Gas

What I knew

When there is a chemical change gas



When there is a physical change there gas is released.



What I learned

Bottle #1 had a



change.

Bottle #2 had a



change.









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Pg 6